

January 8, 2016
Prelim Exam – Day 2

Begin each problem on a separate sheet of paper. Write your **Prelim ID #** in the top right-hand corner of each and every page you submit.

Problem 1:

A particle of mass m in the infinite square well (of width a) starts out in the left half of the well, and is (at $t = 0$) equally likely to be found at any point in that region.

(a) What is its initial wave function, $\Psi(x, 0)$? (Assume it is real. Don't forget to normalize it.)

(b) What is the probability that a measurement of the energy would yield the value $\pi^2 \hbar^2 / 2ma^2$?

Problem 2:

A particle of mass m is in the state

$$\Psi(x,t) = A \exp\{-a[(mx^2/\hbar) + it]\}$$

where A and a are positive real constants.

- (a) Find A .
- (b) For what potential energy function $V(x)$ does Ψ satisfy the Schrödinger equation?
- (c) Calculate the expectation values of x , x^2 , p , and p^2 .

Problem 3:

In the Earth's reference frame, a tree is at the origin and a pole is at $x = 30$ km. Lightning strikes both the tree and the pole at $t = 10 \mu\text{s}$. The lightning strikes are observed by a rocket traveling in the x -direction at $0.5c$.

- (a) What are the spacetime coordinates for these two events in the rocket's reference frame?
- (b) Are the events simultaneous in the rocket's frame? If not, which occurs first?

Problem 4:

Consider a head-on, elastic collision between a massless photon (momentum \mathbf{p}_0 and energy E_0) and a stationary free electron.

(a) Assuming that the photon bounces directly back with momentum \mathbf{p} (in the direction of $-\mathbf{p}_0$) and energy E , use conservation of energy and momentum to find p .

(b) Verify that your answer agrees with that given by Compton's formula with $\theta = \pi$.

Problem 5:



Figure Caption: Shown above is the energy level diagram for a hydrogen atom, showing the four lowest energy levels. Ignoring electron spin s , there is one independent state with energy $= -13.6$ eV, four independent states with energy $= -3.4$ eV, nine independent states with energy $= -1.5$ eV, etc.

Estimate the ratio of the probability that a hydrogen atom at warm room temperature (300 K) is in one of its first excited states ($n = 2$) relative to the probability of it being in the ground state ($n = 1$). Don't forget to take degeneracy into account. Show all steps in the calculation, even if you think that the answer will be trivial. The value of Boltzmann's constant, $k = 8.62 \times 10^{-5}$ eV/K.