

PHYS 1311 Chapter 8 EOC Problems

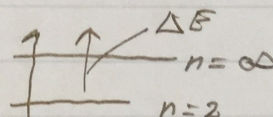
Problems P9, P11, P18, P23, P24, P26 4-20-2016

Problem P9 $E_n = -\frac{13.6 \text{ eV}}{n^2}$

$n=2 \rightarrow n=\infty$

$$\Delta E = |E_\infty - E_2| = \left| -\frac{13.6}{\infty^2} - \left(-\frac{13.6}{2^2} \right) \right|$$

$$= \left| 0 + \frac{13.6}{4} \right| = \boxed{3.4 \text{ eV}}$$



Problem P11

$N_0 = 10,000 \text{ atoms}$

$\tau \equiv \text{lifetime} = 25 \times 10^{-9} \text{ s}$

$t_i = 0, t_f = 12 \times 10^{-9} \text{ s} = t$

$$N(t) = N_0 \exp(-t/\tau) = 10,000 \exp(-12/25)$$

$$= \boxed{6190 \text{ atoms}}$$

Problem P18

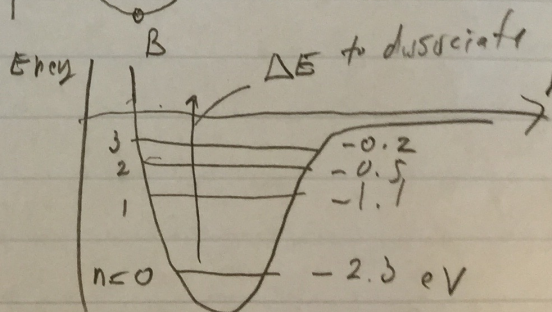
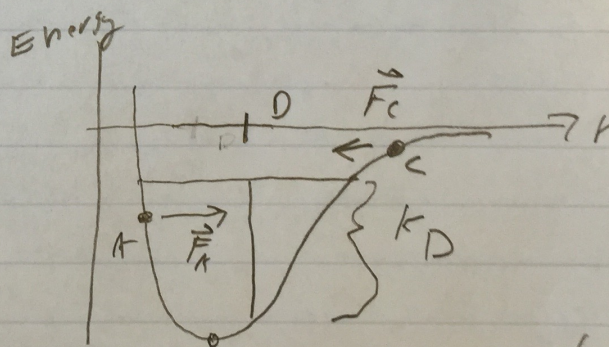
- a) $\vec{F}_A \rightarrow \hat{r} \text{ direction}$
 $\vec{F}_C \rightarrow -\hat{r} \text{ direction}$
 $\vec{F}_B = 0$

since $F_r = -\frac{dU}{dr}$

$|\vec{F}_A| > |\vec{F}_C| > |\vec{F}_B|$

b) To dissociate

$\Delta E > \boxed{2.3 \text{ eV}}$



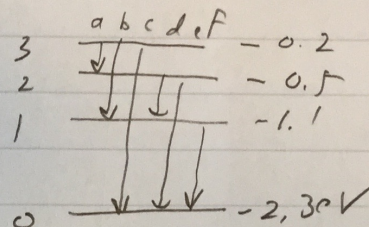
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Problem P18 (cont'd)

c) Photon energies are:

$$E_{ph a} = 0.3 \text{ eV}, E_{ph b} = 0.9 \text{ eV}$$



$$E_{ph c} = 2.1 \text{ eV}, E_{ph d} = 0.6 \text{ eV}, E_{ph e} = 1.8 \text{ eV}, E_{ph f} = 1.2 \text{ eV}$$

d) Total energy = $E = E_{rest} + K + U = Mc^2$

so $n \neq 0$, and $n=1$

$$E_0 = mc^2 + K_0 + U_0, E_1 = Mc^2 + K_1 + U_1$$

$$= M_1 c^2 = M_2 c^2$$

↑ initial mass

$$\Delta E = E_1 - E_0 = K_1 - K_0 + U_1 - U_0 = \Delta M c^2$$

$$\text{or } \Delta M = \frac{\Delta K + \Delta U}{c^2} > 0 \quad [\text{here } 1.2 \text{ eV}/c^2]$$

• The higher excited state has more internal energy and therefore more inertial mass

Problem P23

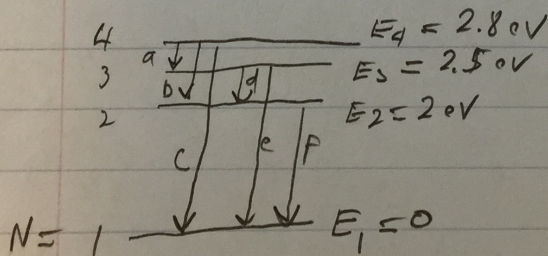
observed

$$E_{ph} = 0.3, 0.5, 0.8, 2, 2.5, 2.8 \text{ eV}$$

$$E_a = E_d, E_b = E_f$$

$$E_c = E_e$$

a) Try the following energy level diagram

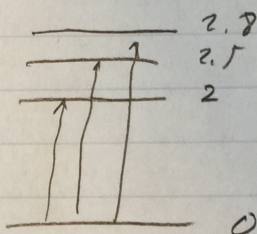


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b) Since the energy intervals are not equally spaced, the harmonic oscillator would not be a good model for this system.

c) Absorption photos

$$E = 2, 2.5, 2.8 \text{ eV}$$



→ other energy diagrams possible

Problem P26 HCl molecule,

a) For harmonic oscillator,

$$E_n = n\hbar\omega_0 + E_0$$

$$m = m_H + m_{Cl}$$

$$= 1u + 36u$$

$$= 37u$$

$$= 37(1.7 \times 10^{-27} \text{ kg})$$

$$\omega_0 = \sqrt{\frac{k_s}{m}}, \text{ estimate } k_s \sim 10 \text{ N/m}$$

$$= \sqrt{\frac{10 \text{ N/m}}{37(1.7 \times 10^{-27} \text{ kg})}} = 1.3 \times 10^{13} \text{ rad/s}$$

$$\hbar\omega_0 = \left(\frac{6.6 \times 10^{-34} \text{ J}\cdot\text{s}}{2\pi} \right) \times \frac{1 \text{ eV}}{1.602 \times 10^{-19} \text{ J}} \left(1.3 \times 10^{13} \text{ rad/s} \right)$$

$$\approx \boxed{0.01 \text{ eV}}$$

b) $\Delta E_{1 \rightarrow 0} = 0.01 \text{ eV}$

$\Delta E_{2 \rightarrow 0} = 0.02 \text{ eV}, \text{ etc.}$

$$\lambda = \frac{hc}{\Delta E} = \frac{12398 \text{ eV}\cdot\text{\AA}}{0.01 \text{ eV}}$$

$$= 1.2 \times 10^6 \text{\AA} \approx 100 \text{ }\mu\text{m}$$

$$\Rightarrow \boxed{\text{infrared}}$$